

Engineering And Chemical Thermodynamics

Koretsky Solutions

Volatility (chemistry) John Wiley & Sons. (2013). ISBN 978-1-119-17764-7. 279-281. Sons. A highly volatile substance such as rubbing alcohol (isopropyl alcohol) will quickly evaporate, while a substance with low volatility such as vegetable oil will remain. At a given temperature and pressure, a substance with high volatility is more likely to exist as a vapour, while a substance with low volatility is more likely to be a liquid or solid. Zumdahl, In chemistry, volatility is a material quality which describes how readily a substance vaporizes. pp. pp. Differences in volatility can be observed by comparing how fast substances within a group evaporate (or sublime in the case of solids) when exposed to the atmosphere. Engineering and Chemical Thermodynamics. 639-641. Volatility can also describe the tendency of a vapor to condense into a liquid or solid; less volatile substances will more readily condense from a vapor than highly volatile ones. Koretsky, Milo D f4616a9b91

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